


Solutions to Exercises: Organic Chemistry

Class X Chapter 12 – Organic Chemistry 
Chemistry

Book Name: Selina Concise

EXERCISE 12 A

Solution 1:

(a) 2,2-dimethylpropane
(b) 2-methyl butane
(c) Prop-1-ene
(d) 2,2-dimethyl pentane
(e) Pent-2-yne
(f) 3-methyl but-1-yne
(g) 2,3-dichloropentane
(h) 3-methylheptane
(i) 2-methyl butane
(j) Hept-2-yne
(k) 2,2-dimethyl hexanal
(l) Pentan-2-ol
(m) 4-methylpentanoic acid
(n) 2-bromo-2-methyl butane
(o) 1-bromo-3-methyl butane

Solution 2:
The structure of the following compounds are:

(a) Prop-1-ene
 $\text{CH}_3 - \text{CH} = \text{CH}_2$

(b) 2,3-dimethylbutane
 $\text{CH}_3 - \text{CH}(\text{CH}_3) - \text{CH}(\text{CH}_3) - \text{CH}_3$

(c) 2-methylpropane
 $\text{CH}_3 - \text{C}(\text{CH}_3)_3$

(d) 3-hexene
 $\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH} - \text{CH}_2 - \text{CH}_3$

(e) Prop-1-yne
 $\text{CH}_3 - \text{C}\equiv\text{CH}$

(f) 2-methylprop-1-ene
 $\text{CH}_3 - \text{C}(\text{CH}_3) = \text{CH}_2$

(g) Alcohol with molecular formula $\text{C}_4\text{H}_{10}\text{O}$
 $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{OH}$

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